

A New Route to Substituted 3-Methoxycarbonyldihydropyrans; Enantioselective Synthesis of (–)-Methyl Elenolate

Susumi Hatakeyama, Noriko Ochi, Hirotochi Numata, and Seiichi Takano*

Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980, Japan

A stereoselective chiral synthesis of (–)-methyl elenolate has been achieved by employing a newly developed method for the construction of substituted 3-methoxycarbonyldihydropyrans based on a radical cyclisation reaction.

A number of secoiridoid monoterpenes and biogenetically related heteroyohimbine indole alkaloids possess characteristic highly substituted 3-methoxycarbonyldihydropyran ring systems, as exemplified by elenolic acid (**1**), sarracenin (**3**), and ajmalicine (**4**). The stereoselective assembly of these ring systems is one of the most crucial problems in the synthesis of this family of compounds.^{1,2} For the purpose of developing an effective method for the construction of substituted 3-methoxycarbonyldihydropyrans, we chose methyl elenolate (**2**)³ as a synthetic target. Methyl elenolate (**2**) is the methyl ester of elenolic acid (**1**), a secoiridoid monoterpene isolated from olive (*Oleau europea*).⁴ The combination of its broad-range antiviral activity⁵ and its synthetic utility^{3a} as a precursor of ajmalicine (**4**), a therapeutically important heteroyohimbine indole alkaloid, made (**2**) an attractive target. We now report an enantio- and stereo-selective synthesis of (–)-methyl elenolate (**2**) by use of a new methodology based on a radical cyclisation reaction.⁶

Initially, our strategy was tested in a model study as follows. Thus, five bromoacetals (**6a–e**)[†] were prepared from the

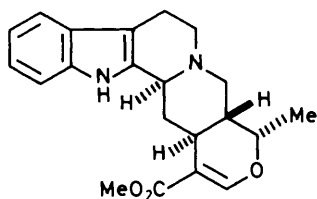
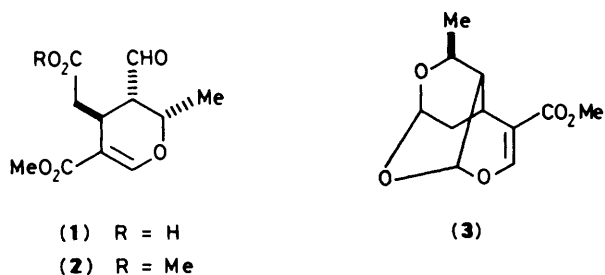
corresponding δ -hydroxy α,β -unsaturated esters (**5a–e**) by addition⁷ of methyl propiolate followed by treatment with *N*-bromosuccinimide (NBS) in methanol. Reaction⁸ of (**6a–e**) with tributyltin hydride in the presence of a catalytic amount of azobisisobutyronitrile (AIBN) in boiling benzene followed by treatment with acid in the same flask led to formation of the 3-methoxycarbonyldihydropyran derivatives (**7a–e**). The yield and the stereochemical outcome of this reaction sequence are summarized in Table 1.

Table 1. Synthesis of 3-methoxycarbonyldihydropyrans (**7a–e**).

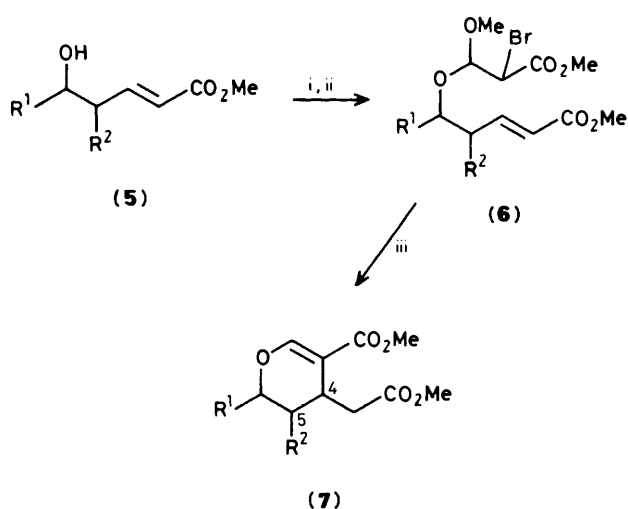
(5)	% Yield of (6)	% Yield of (7) (<i>syn</i> : <i>anti</i>) ^a
a ; R ¹ = H, R ² = H	86	80
b ; R ¹ = Me, R ² = H	83	74 (3 : 1)
c ; R ¹ = H, R ² = Me	89	74 (1 : 3)
d ; R ¹ = Me, R ² = Me (<i>anti</i>)	94	64 (1 : 2) ^b
e ; R ¹ = Me, R ² = Me (<i>syn</i>)	94	80 (1 : 3) ^b

^a Determined by 500 MHz ¹H n.m.r. ^b Relative configuration between C-4 and C-5.

[†] All new compounds exhibited satisfactory spectral (¹H n.m.r., i.r., and high resolution mass) data.



(4)
Scheme 1

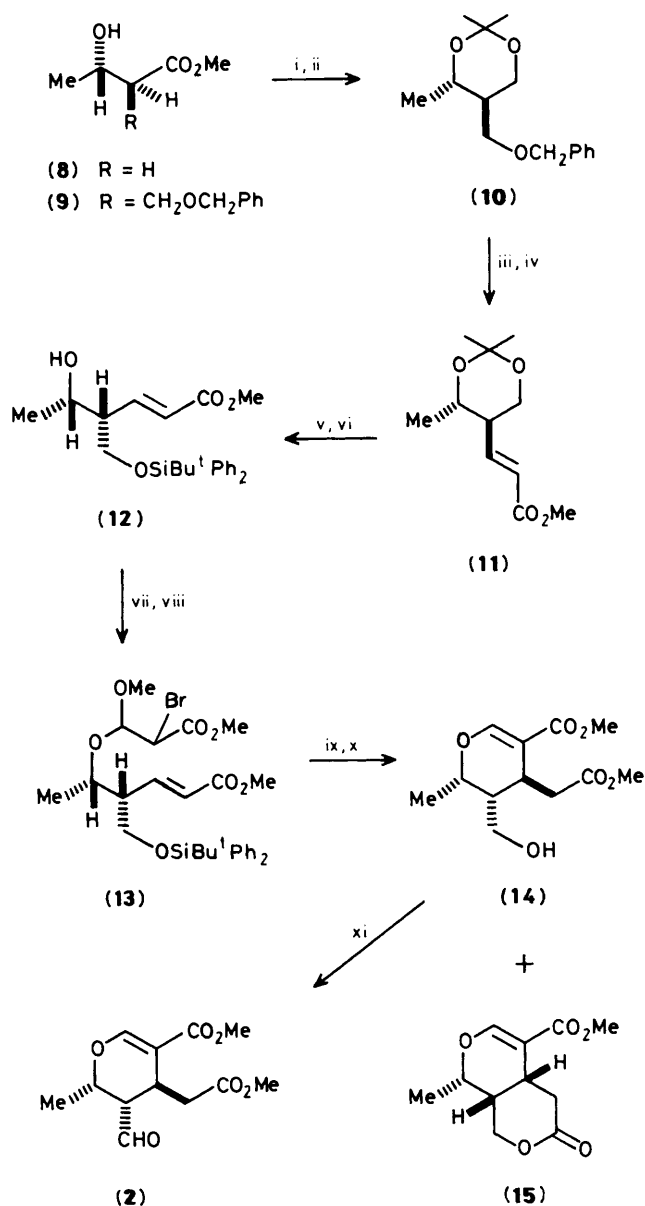


Scheme 2. Reagents and conditions: i, methyl propiolate (3.5 equiv.), *N*-methylmorpholine (26 mol%), toluene; ii, NBS (1.2 equiv.), NaHCO₃ (1.3 equiv.), MeOH; iii, Bu₃SnH (1.2 equiv.), AIBN (5 mol%), benzene, reflux (1 h), then add TsOH (40 mol%), reflux (1 h).

Having established a new method for the construction of substituted 3-methoxycarbonyldihydropyrans, we investigated the synthesis of (2). The hydroxy ester (9),⁹ prepared from commercially available (*S*)-methyl 3-hydroxybutyrate (8) (78% enantiomeric excess)[‡] by stereoselective alkylation, was subjected successively to reduction and protection to give the acetonide (10), [α]_D²⁵ -16.8° (*c* 1.01, CHCl₃),§ in 70%

[‡] The optical purity of (8), purchased from Wako Pure Chemical Industries Ltd., was determined on the basis of the specific rotation, $\alpha_D^{24} + 18.4^\circ$ (neat) [enantiomerically pure enantiomer:¹¹ $\alpha_D^{22} - 23.5^\circ$ (neat)]. This value was consistent with that determined by 500 MHz ¹H n.m.r. analysis of both *R*- and *S*-methoxy(trifluoromethyl)phenylacetyl ester derivatives of (9).

§ All the optical rotations reported here should be regarded as corresponding to 78% enantiomeric excess.



Scheme 3. Reagents and conditions: i, LiAlH₄, THF; ii, Me₂C(OMe)₂, TsOH, acetone; iii, Li, liq. NH₃/THF, -33°C; iv, (COCl)₂/Me₂SO/Et₃N, CH₂Cl₂, -50 to 25°C, then add Ph₃P=CHCO₂Me; v, 1M HCl/THF (1:1); vi, Ph₂Bu⁺SiCl, imidazole, CH₂Cl₂; vii, methyl propiolate (3.5 equiv.), *N*-methylmorpholine (26 mol%), toluene; viii, NBS (1.2 equiv.), NaHCO₃ (1.3 equiv.), MeOH; ix, Bu₃SnH (1.2 equiv.), AIBN (5 mol%), benzene, reflux (1 h), then add TsOH (40 mol%), reflux (1 h); x, 46% aq. HF/MeOH (1:3); xi, (COCl)₂/Me₂SO/Et₃N, CH₂Cl₂, -50 to 25°C.

yield. Upon sequential debenzoylation, Swern oxidation, and Wittig reaction,¹⁰ the α,β -unsaturated ester (11), [α]_D¹⁹ -28.6° (*c* 1.01, CHCl₃), was obtained in 95% overall yield from (10). Acid hydrolysis followed by selective silylation afforded the *O*-silyl derivative (12),[¶] [α]_D²³ -9.0° (*c* 0.98, CHCl₃), in 92% yield. Compound (12) was converted into the

[¶] The structure of (12) was confirmed by 90 MHz ¹H n.m.r. analysis of the corresponding acetate, δ (CDCl₃) 1.06 (9H, s), 1.16 (3H, d, *J* 6.8 Hz), 1.97 (3H, s), 2.62 (1H, m), 3.70 (2H, d, *J* 6.5 Hz), 3.77 (3H, s), 5.15 (1H, quint, *J* 6.8 Hz), 5.90 (1H, dd, *J* 15.5 and 1.1 Hz), 6.91 (1H, dd, *J* 15.5 and 8.5 Hz), 7.27-7.80 (10H, m).

bromo acetal (**13**) in 89% yield in a manner similar to that described for the model study. Upon radical cyclisation followed by acid treatment in the same flask and desilylation, (**13**) yielded the alcohol (**14**), $[\alpha]_D^{24} -62^\circ$ (c 0.89, CHCl_3), and the lactone (**15**) in the ratio 4:1, in 58% yield. Finally, Swern oxidation of (**14**) furnished (–)-methyl elenolate (**2**) in 88% yield. The synthetic substance exhibited spectral properties (500 MHz ^1H n.m.r., i.r., and mass) in accord with those of an authentic sample.^{3b} The optical rotation, $[\alpha]_D^{23} -94.7^\circ$ (c 0.76, CHCl_3) [lit.,^{3c} -121° (c 0.68, CHCl_3)], shows that there is no racemisation step in the present synthetic route, since the optical purity of (**2**) was almost identical with that of the starting ester (**8**). Purification of (**2**) (h.p.l.c. on octadecylsilane) allowed us to obtain a pure sample without any epimerisation at the position α to the formyl group.||

Received, 27th April 1988; Com. 8/01659C

References

1 L.-F. Tietze, *Angew. Chem., Int. Ed. Engl.*, 1983, **22**, 828, and references cited therein.

|| Column chromatography (SiO_2 ; 1:1 ether–hexane) brought about epimerisation, to give a 88:12 mixture of (**2**) and its epimer, respectively.

- 2 S. Takano, K. Morikawa, and S. Hatakeyama, *Tetrahedron Lett.*, 1983, **24**, 401; S. Takano, S. Satoh, and K. Ogasawara, *J. Chem. Soc., Chem. Commun.*, 1988, 59.
- 3 For the synthesis of (\pm)-methyl elenolate, see (a) R. C. Kelly and I. Schletter, *J. Am. Chem. Soc.*, 1973, **95**, 7156; for the synthesis of (–)-methyl elenolate, see (b) S. Hatakeyama, K. Saijo, and S. Takano, *Tetrahedron Lett.*, 1985, **26**, 865; (c) S. Satoh, Ph.D. Dissertation, Tohoku University, Sendai, Japan, 1988.
- 4 W. L. C. Weer, V. Gerris, J. E. Ribbers, P. J. Van Ree, H. C. Beyerman, and J. S. Boniekoe, *Recl. Trav. Chim. Pays-Bas.*, 1957, **76**, 839.
- 5 H. E. Renis, *Antimicrob. Agents Chemother.*, 1970, **1969**, 167; M. G. Soret, *ibid.*, 1970, **1969**, 160; E. N. DeYoung, *ibid.*, 1970, **1964**, 173.
- 6 For reviews on radical reactions in organic synthesis, see M. Ramaiah, *Tetrahedron*, 1987, **43**, 3541; B. Giese, 'Radicals in Organic Synthesis: Formation of Carbon–Carbon Bonds,' ed. J. E. Baldwin, Pergamon, New York, 1986, vol. 5.
- 7 A. A. L. Gunatilaka, H. Hirai, and D. G. I. Kingston, *Tetrahedron Lett.*, 1983, **24**, 5457.
- 8 Y. Ueno, K. Chino, M. Watanabe, O. Moriya, and M. Okawara, *J. Am. Chem. Soc.*, 1982, **104**, 5564; G. Stork, R. Mook, S. C. Biller, and S. D. Pychnovsky, *ibid.*, 1983, **105**, 3741.
- 9 For the synthesis of the enantiomer, see R. E. Ireland and R. B. Wardle, *J. Org. Chem.*, 1987, **52**, 1780.
- 10 R. E. Ireland and D. W. Norbeck, *J. Org. Chem.*, 1985, **50**, 2198.
- 11 D. Seebach and M. Zueger, *Helv. Chim. Acta*, 1982, **65**, 495.